

Section A (36 marks)

- 1 Express $\cos \theta - 3 \sin \theta$ in the form $R \cos(\theta + \alpha)$, where $R > 0$ and $0 < \alpha < \frac{\pi}{2}$.

Hence show that the equation $\cos \theta - 3 \sin \theta = 4$ has no solution. [6]

- 2 Given that $\left(1 + \frac{x}{p}\right)^q = 1 - x + \frac{3}{4}x^2 + \dots$, find p and q , and state the set of values of x for which the expansion is valid. [7]

- 3 Fig. 3 shows the curve $y = x^4$ and the line $y = 4$.

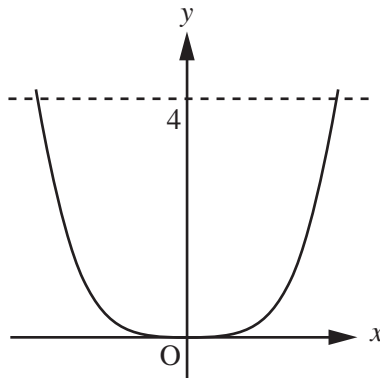


Fig. 3

The finite region enclosed by the curve and the line is rotated through 180° about the y -axis. Find the exact volume of revolution generated. [4]

- 4 Solve the equation $2 \sin 2\theta = 1 + \cos 2\theta$ for $0^\circ \leq \theta \leq 180^\circ$. [5]

- 5 In Fig. 5, triangles ABC, ACD and ADE are all right-angled, and angles BAC, CAD and DAE are all θ .

$AB = x$ and $AE = 2x$.

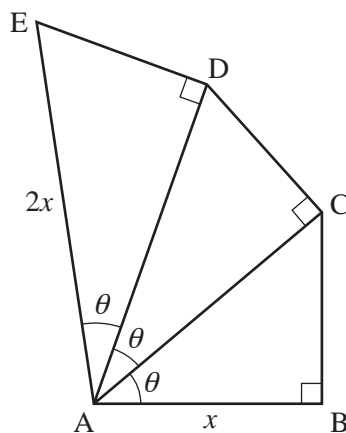


Fig. 5

- (i) Show that $\sec^3 \theta = 2$. [3]

- (ii) Hence show the ratio of lengths ED to CB is $2^{\frac{2}{3}} : 1$. [4]

- 6 P is a general point on the curve with parametric equations $x = 2t$, $y = \frac{2}{t}$. This is shown in Fig. 6. The tangent at P intersects the x - and y -axes at the points Q and R respectively.

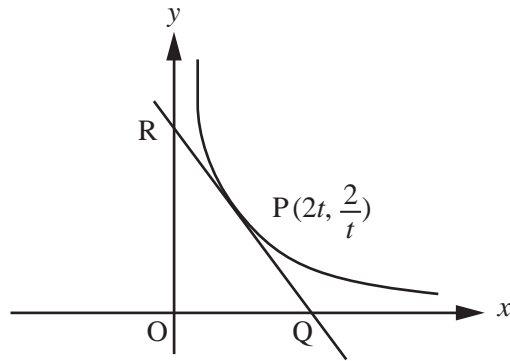


Fig. 6

Show that the area of the triangle OQR, where O is the origin, is independent of t .

[7]

Section B (36 marks)

- 7 Fig. 7 shows a cuboid OABCDEFG with coordinates as shown. The point P has coordinates (4, 2, 0).

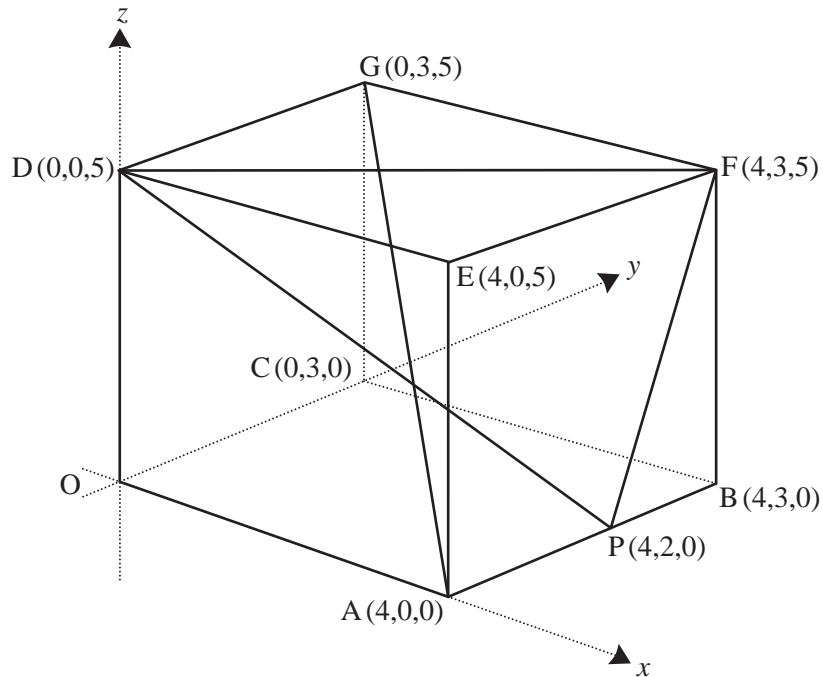


Fig. 7

- (i) Find the length of the diagonal AG. [2]
- (ii) Show that the vector $\mathbf{n} = 15\mathbf{i} - 20\mathbf{j} + 4\mathbf{k}$ is normal to the plane DPF. Hence find the cartesian equation of this plane. [6]

The diagonal AG intersects the plane DPF at Q.

- (iii) Write down a vector equation of the line AG. Hence find the coordinates of the point Q, and the ratio AQ:QG. [6]
- (iv) Find the acute angle between the line AG and the plane DPF. [4]

8 (i) Show that $\frac{1}{2+x} + \frac{1}{2-x} = \frac{4}{(2+x)(2-x)}$. [1]

In a chemical reaction, the time t minutes taken for a mass x mg of a substance to be produced is modelled by the equation

$$t = \ln\left(\frac{2+x}{2-x}\right).$$

(ii) Show that when $t = 0$, $x = 0$. [2]

(iii) Show that the rate of change of x is proportional to the product of $(2 + x)$ and $(2 - x)$, and find the constant of proportionality. [4]

(iv) Show that $x = \frac{2(1 - e^{-t})}{1 + e^{-t}}$.

Hence determine the long-term mass of the substance predicted by this model. [4]

In another chemical reaction, the mass x mg at time t minutes is modelled by the differential equation

$$\frac{dx}{dt} = k(2+x)(2-x)e^{-t},$$

where k is a positive constant, and $x = 0$ when $t = 0$.

(v) Show by integration that, for this reaction, $\ln\left(\frac{2+x}{2-x}\right) = 4k(1 - e^{-t})$. [5]

(vi) Given that the long-term mass of this substance is 1.85 mg, find the value of k . [2]

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